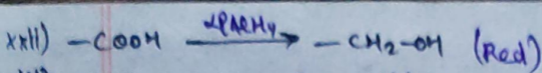


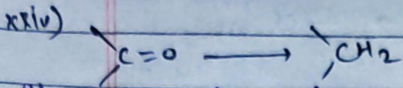
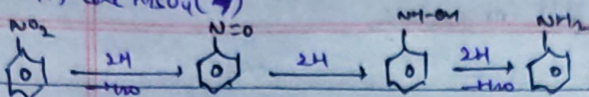
All Conversions

- i)  $-Cl \xrightarrow{aq. KOH} -OH$  ( $S_N2$ )  
 ii)  $-OH \longrightarrow -Cl$   
 i)  $PCl_5$  ( $S_N2$ )  
 ii)  $PCl_3$  ( $S_N2$ )  
 iii)  $ZnCl_2/HCl$  ( $S_Ni$ )  
 iv)  $SOCl_2$  ( $S_Ni$ )  
 v)  $SOCl_2$  (py) ( $S_N2$ )
- iii)  $-Cl \xrightarrow{K^+ CN^-} -C \equiv N$  ( $S_N2$ )  
 iv)  $-Cl \xrightarrow{AgCN} -N \equiv C$  ( $S_N2$ )  
 v)  $\begin{array}{c} | & | \\ -C & -C- \\ | & | \end{array} \longrightarrow \begin{array}{c} \diagup & \diagdown \\ C & = & C \\ \diagdown & \diagup \end{array}$  (i) alc KOH ( $E_2$ )  
 (ii)  $H_2SO_4$  ( $E_1$ )  
 vi)  $\begin{array}{c} | & | \\ -C & -C-OH \\ | & | \end{array} \longrightarrow \begin{array}{c} \diagup & \diagdown \\ C & = & C \\ \diagdown & \diagup \end{array}$   
 (i)  $H_2SO_4$  or  $H_3PO_4$  /  $DEI$  (ii)  $Al_2O_3, P_2O_5, POCl_3$  ( $E_2$ )  
 (iii)  $ThO_2$  ( $E_1cb$ )  
 $\Delta$
- vii)  $-C \equiv N \longrightarrow -CH_2-NH_2$   
 (i)  $2H_2 / Pd$  (ii)  $LiAlH_4$
- viii)  $-C \equiv N \xrightarrow[Full Hydro]{2H_2O/\Delta} -C-OOH$   
 ix)  $-C \equiv N \xrightarrow[Partial]{H_2O} -CONH_2$   
 x)  $-CONH_2 \longrightarrow -CH_2NH_2$  (i)  $H_2/Pd$   
 (ii)  $LiAlH_4$
- xi)  $-CONH_2 \xrightarrow[Reduction]{Br_2/KOH} -NH_2$  (Hofmann Bromide Degradation)
- xii)  $R-NH_2 \xrightarrow[Nitrosation]{HNO_2} R-OH$  (Diazotization)
- xiii)  $Ar-NH_2 \xrightarrow[1) HNO_2 (0-5^\circ), 2) warm H_2O]{Ar-OH}$  (Diazotization)  
 (i)  $Cu_2O_2$  (Sandmeyer)  
 (ii)  $Cu/HCl$  (Gattermann)
- xiv)  $-N_2Cl \longrightarrow -Cl$   
 (i)  $Cu_2Br_2$   
 (ii)  $Cu/HBr$
- xv)  $-N_2Cl \longrightarrow -Br$   
 (i)  $Cu_2Br_2$   
 (ii)  $Cu/HBr$
- xvi)  $-N_2Cl \xrightarrow[1)  $NABH_4$ , 2)  $HBF_4$ ]{-F}$  (Baltz Scheinman)
- xvii)  $-N_2Cl \xrightarrow{Cu(CN)_2} -CN$  (Sandmeyer)
- xviii)  $-CHO \xrightarrow{[O]} -COOH$   
 i)  $KMnO_4$  ( $H^+$ ) (ii)  $K_2Cr_2O_7$  ( $H^+$ ) (iii)  $CrO_3$  ( $H_2SO_4$ ) Acetone  
 (iv)  $HNO_3$  (Jones)
- v) Tollen's vi) Fehling's vii) Benedict's
- xix)  $-CHO \xrightarrow{2[H]} -CH_2OH$   
 i)  $H_2/Pd$  ii)  $LiAlH_4$  iii)  $SBH$
- xx)  $-CH_2OH \longrightarrow -CHO$   
 i)  $PCC$  ii)  $PDC$  iii) Collins iv)  $Cu/300^\circ$   
 v)  $MnO_2$  (Allylic) vi)  $NBS$
- xxi)  $-CH_2OH \longrightarrow -COOH$   
 i)  $KMnO_4$  ( $H^+$ ) ii)  $CrO_3 + H_2SO_4$  (Acetone) (iii)  $K_2Cr_2O_7$



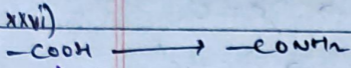
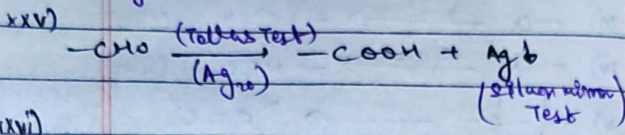
i) Sn/HCl ii) H<sub>2</sub>Pd iii) Fe/HCl iv) LAH

(\*) conc H<sub>2</sub>SO<sub>4</sub> (4)

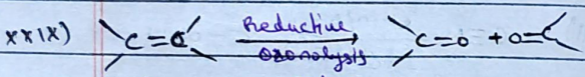
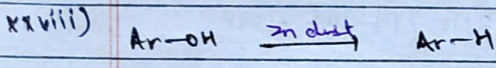
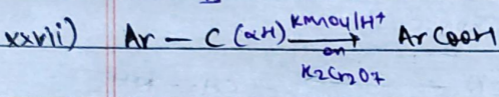
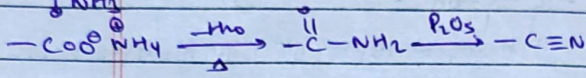


(i) NH<sub>2</sub>-NH<sub>2</sub> (KOH) (Wolf Kishner) (Base) \*

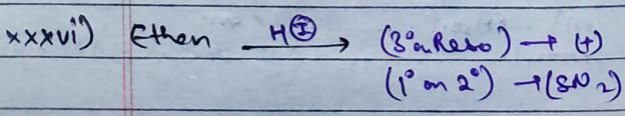
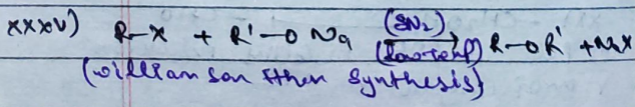
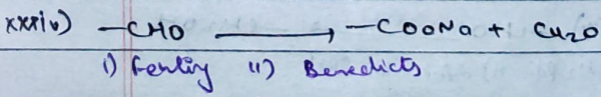
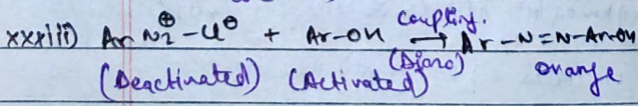
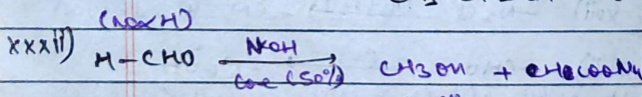
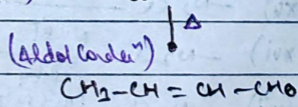
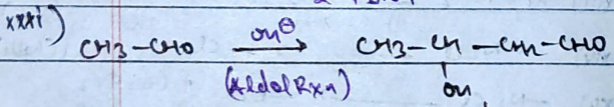
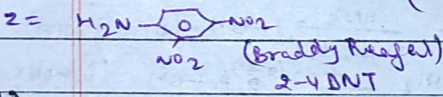
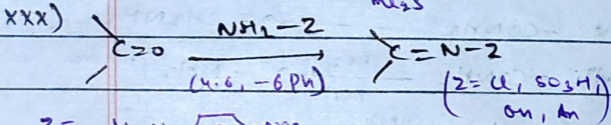
(ii) Zn/Hg (HCl) (Clemmensen Redn) (Acid) \*

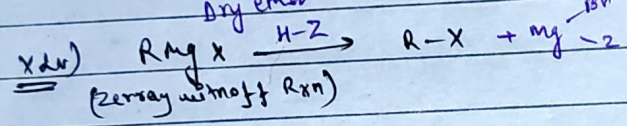
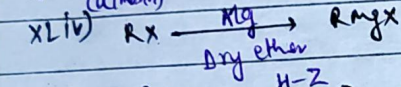
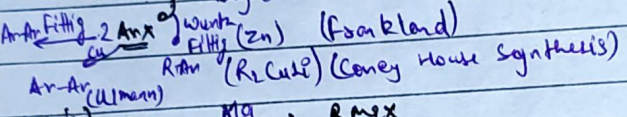
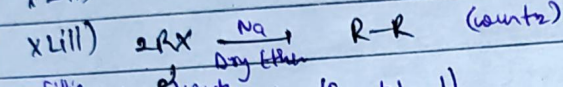
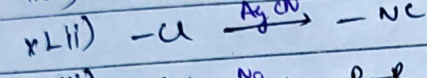
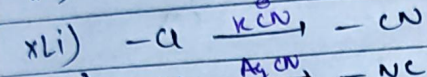
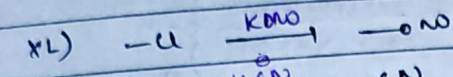
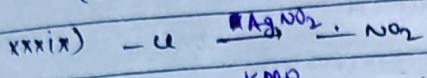
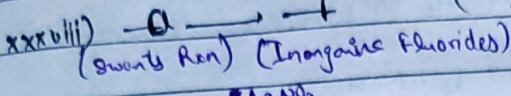
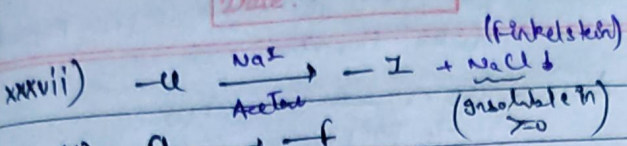


↓ NH<sub>3</sub>



O<sub>3</sub> | Zn/HCl | PPh<sub>3</sub>  
on  
meq





C → SP Hybr. Carbon

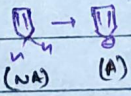
O → H attached to (O)

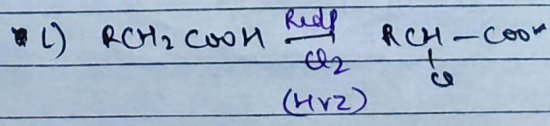
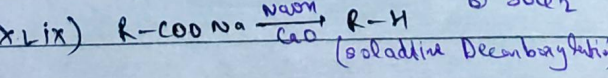
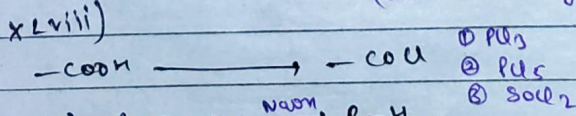
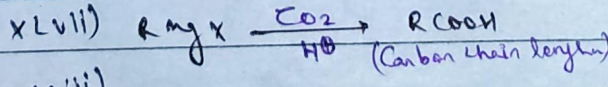
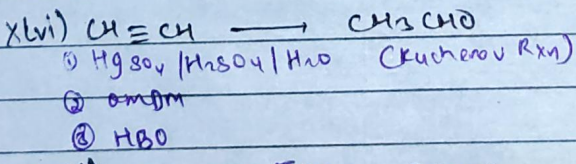
N → H attached to (N)

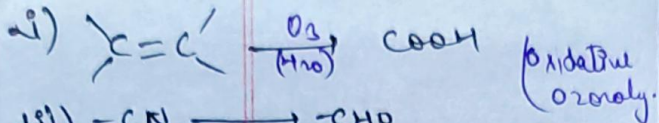
S → H attached to (S)

N → H attached to (NO<sub>2</sub>)

A → Active methylene group  $CH_2^{(2)}$   
(Z = EWG)

Q → Quasi Aromatic 





(i)  $\text{SnCl}_2 / \text{HCl}$  (ii)  $\text{Hno} \rightarrow$  Stephen method

(iii)  $\text{DIBAL-H, Hno}$



(i)  $\text{H}_2 / \text{Pd} / \text{BaSO}_4$  (Rosenmund)  
xylene

(ii)  $\text{DIBALH}$

